

Ring C Homologation of Aporphines. A New Synthesis of Homoaporphines

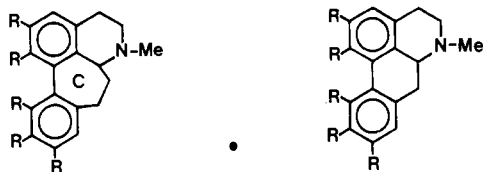
José L. Castro, Luis Castedo, and Ricardo Riguera*

Departamento de Química Orgánica y Sección de Alcaloides del CSIC, Facultad de Química, Universidad de Santiago de Compostela, Santiago de Compostela, Spain

Received October 31, 1986

We describe an easy method for the ring C homologation of aporphines that is based on dichlorocarbene addition to dehydroaporphinoids and phenanthrene derivatives. 17a and 17d, however, gave the insertion products 23 and 21b, respectively.

Homoaporphines constitute a small group of isoquinoline alkaloids closely related to aporphines. The main structural difference between them is the existence of a seven-membered ring C in homoaporphines.²



HOMOAPORPHINES

APORPHINES

Up to the present, the preparation of homoaporphines has been performed by ab initio synthesis. All these methods invariably pass through the required tetrahydrophenethylisoquinoline, which by phenolic or non-phenolic coupling,³ by photochemical cyclization,⁴ or through quinol acetates⁵ gives the desired homoaporphine. Curiously, although aporphines are the largest group of isoquinoline alkaloids, no efforts have been made toward their conversion into homoaporphines. We present here an easy and efficient strategy that covers this subject and that is based on dichlorocarbene addition to dehydroaporphinoids or phenanthrene derivatives.

Results and Discussion

Dehydroaporphine Approach. The enamine character of dehydroaporphines is well-known and explains their reactivity at C-7 toward acids⁶ and dichlorocarbene.⁷ We reasoned that by blocking the nitrogen lone electron pair the reactivity of the dehydroaporphine should be different, particularly the olefinic character of the C-6a-C-7 double bond. Thus, when the N-protected dehydroaporphine 1 was treated with dichlorocarbene generated by the phase-transfer method (PTM),⁸ we obtained the adduct 2 in 86% yield after 20 min of stirring at room temperature. Extended reaction times and/or heating gave

complex reaction mixtures without practical use. Reaction of 2 with LiAlH₄ in refluxing THF produced a mixture of 7-chloro-7,8-didehydrohomodirectrine (6) and 7,8-didehydrohomodirectrine (5) in 77% and 3% yields, respectively; this conversion involves the reduction of the carboxyethyl function of 2 to an N-methyl group, followed by ring expansion with the aid of the nitrogen lone pair and subsequent reduction of the iminium cation (Figure 1). Catalytic hydrogenation of 6 (10% Pd-C, NaOAc, EtOH) gave homodirectrine (7) in 63% overall yield with respect to 1.

Phenanthrene Approach. The above ring C homologation of aporphines starts from an N-carboxy derivative, whose preparation from aporphines⁹ or by total synthesis¹⁰ requires several steps. To reduce this problem, we developed an alternative route based on the addition of dichlorocarbene to N-protected phenanthrene derivatives, which are easily available from aporphines (Figure 2). In this way, treatment of glaucine (8) with trifluoroacetic anhydride gave phenanthrene 9, which upon dichlorocarbene addition (PTM) produced adduct 10. This was easily N-deprotected (MeOH, Na₂CO₃) to amine 11. Regeneration of ring B and homologation of ring C were achieved in one step by thermal electrocyclic ring opening of the dichlorocyclopropane ring of 11 and intramolecular trapping of the developing allylic cation by the nitrogen atom. The 7-chloro-7,8-didehydrohomoglaucine (12) thus obtained was converted into homoglaucine (13) by catalytic hydrogenation (43% overall yield from 8).

The choice of the nitrogen-protecting group was of critical importance. Thus, while a trifluoroacetyl group was found to be ideal for our purposes because it resists Cl₂C: (PTM) and is easily removed, a carboxyethyl group (as in 16) could not be hydrolyzed without the required dichlorocyclopropane being affected.

All our attempts to convert the chlorovinyl unit of 12 to a carbonyl group were unsuccessful. Thus, 12 was inert toward hydrolysis promoted by mercuric acetate,¹¹ palladium(II) chloride,¹² or sulfuric acid at 0 °C.¹¹ This lack of reactivity may be ascribed to the double bond not being activated because the nitrogen is protonated or coordinated to the metal. Under basic conditions (Me₂SO, KOH; 25 °C; 4 days), compound 12 reacted slowly, giving a complex mixture of products. Nevertheless, although 12 was recovered unchanged from its reaction with Zn(Ag),¹³ the chlorovinyl compound 12 could be dehalogenated with

(1) Part of this work has been published in a preliminary form: Castro, J. L.; Castedo, L.; Riguera, R. *Tetrahedron Lett.* 1985, 26, 1561.

(2) (a) Shamma, M. *The Isoquinoline Alkaloids*; Academic: London, 1972; p 469. (b) Shamma, M.; Moniot, J. L. *Isoquinoline Alkaloids Research*; Plenum: London, 1978; p 371.

(3) (a) Kupchan, S. M.; Dhingra, O. P.; Kim, C. *J. Org. Chem.* 1976, 41, 4049. (b) Kupchan, S. M.; Dhingra, O. P.; Kim, C.; Kameswaran, V. *J. Org. Chem.* 1976, 41, 4047. (c) Kametani, T.; Satoh, F.; Yagi, H.; Fukumoto, K. *Chem. Commun.* 1967, 1103. (d) Battersby, A. R.; McDonald, E.; Munro, M. H. G.; Ramage, R. *Chem. Commun.* 1967, 934.

(4) (a) Kametani, T.; Satoh, Y.; Fukumoto, K. *Tetrahedron* 1973, 29, 2027. (b) Kametani, T.; Satoh, Y.; Fukumoto, K. *J. Chem. Soc., Perkin Trans. 1* 1972, 2160.

(5) (a) Hara, H.; Hoshino, O.; Umezawa, B. *J. Chem. Soc., Perkin Trans. 1* 1979, 2657. (b) Hara, H.; Shinoki, H.; Hoshino, O.; Umezawa, B. *Heterocycles* 1983, 20, 2155.

(6) Wenkateswarlu, A.; Cava, M. P. *Tetrahedron* 1976, 32, 2079.

(7) Saá, J. M.; Cava, M. P. *J. Org. Chem.* 1977, 42, 347.

(8) Makosza, M.; Wawrzyniewicz, M. *Tetrahedron Lett.* 1969, 4659.

(9) Castedo, L.; Iglesias, T.; Puga, A.; Saá, J. M.; Suau, R. *Heterocycles* 1981, 15, 915.

(10) Cava, M. P.; Atern, P.; Wakisaka, K. *Tetrahedron* 1973, 29, 2245.

(11) (a) Martin, S. F.; Chou, T. *Tetrahedron Lett.* 1978, 1943. (b) Yoshida, H.; Katasaka, K.; Kobayashi, M.; Matsumoto, T. *Tetrahedron Lett.* 1979, 3489. (c) Arzoumanian, H.; Metzger, J. J. *Organomet. Chem.* 1973, 57, C1.

(12) Julia, M.; Blasioli, C. *Bull. Soc. Chim. Fr.* 1976, 1941.

(13) Clark, R. D.; Heathcock, C. H. *J. Org. Chem.* 1976, 41, 636.

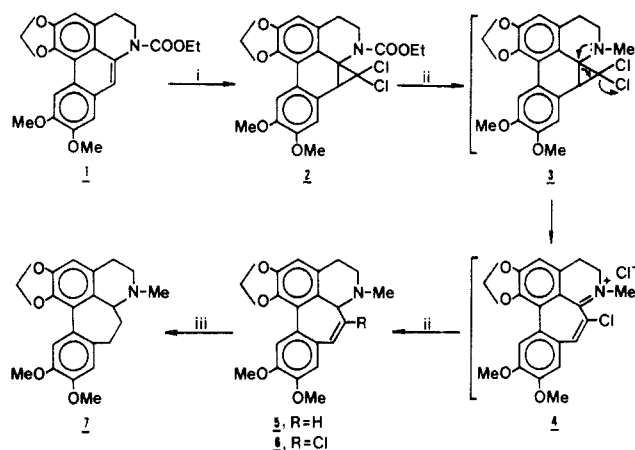


Figure 1. Reagents and conditions: (i) CHCl_3 , 50% NaOH , TBAC, 86%; (ii) LiAlH_4 , THF, 77%; (iii) H_2 , 5% Pd-C, NaOAc, EtOH, 94%.

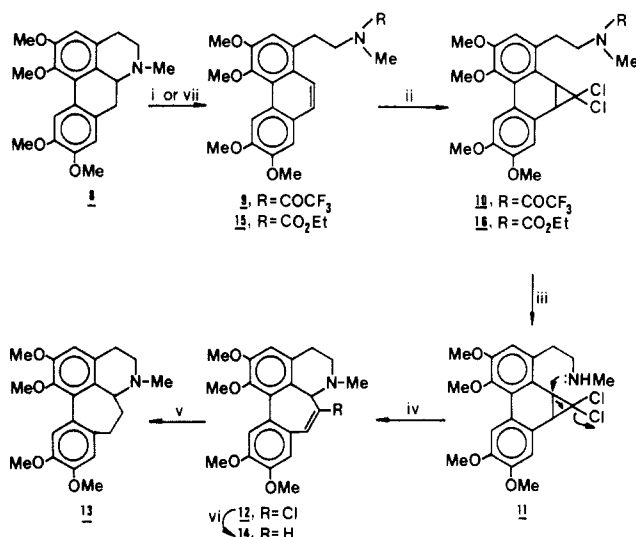


Figure 2. Reagents and conditions: (i) $\text{CF}_3\text{CO}_2\text{O}$, py, 86%; (ii) CHCl_3 , 50% NaOH , TBAC, 69%; (iii) Na_2CO_3 , MeOH, 25 °C; (iv) toluene, reflux, 78% from 10; (v) H_2 , 10% Pd-C, NaOAc, EtOH, 93%; (vi) $\text{LiAlH}_4\text{-AlCl}_3$, THF, 68%; (vii) $(\text{EtOCO})_2\text{O}$, py, 98%.

$\text{LiAlH}_4\text{-AlCl}_3$ (refluxing THF; 46 h), giving 7,8-dehydrohomoglaucine (14) in 63% yield (Figure 2).

Reaction of Dichlorocarbene with C-7-Substituted Dehydroaporphines. In order to extend the above approach to 7-substituted dehydroaporphines, we wanted to know the influence of the C-7 substituent on the enamine character of the dehydroaporphine and, thus, the feasibility of a Cl_2C addition to that double bond.

The poor nucleophilic character of the enamine system of 7-methyl-6a,7-didehydroglaucone (17b) was clearly demonstrated by its slow reaction with methyl iodide and by its inertness toward formaldehyde/ H_2 catalysis.¹⁴ In accordance with this, when 17b was treated with dichlorocarbene as above, the unstable adduct 18a was obtained. Its reaction with LiAlH_4 (or LiAlD_4) in refluxing THF afforded 19 in about 60% yield. Catalytic hydrogenation of 19 (10% Pd-C, EtOH) gave a single diastereomer of 8-methylhomoglaucine (20) in 98% yield (Figure 3). The relative stereochemistry of H-6a and H-8 was deduced by measurement of nuclear Overhauser effects (NOE) and by selective $^1\text{H}\text{-}^1\text{H}$ decoupling experiments on 20a and 20b. The coupling constants for H-6a, H-7 α , H-7 β ,

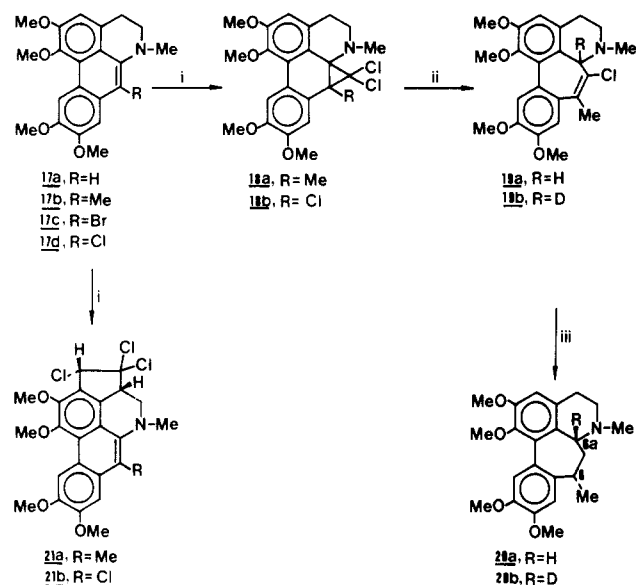


Figure 3. Reagents and conditions: (i) CHCl_3 , 50% NaOH , TBAC; (ii) LiAlH_4 or LiAlD_4 , THF; (iii) H_2 , 10% Pd-C, EtOH.

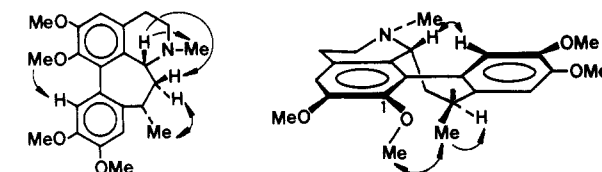


Figure 4. Results of NOE experiments on 20a.

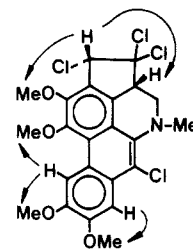


Figure 5. Results of NOE experiments on 21b.

and H-8 were compatible with a cis relationship between H-6a and H-8, and in fact, only this stereochemical arrangement could explain the observed NOE between 8-Me and 1-OMe (Figure 4).

The reaction of 17b with Cl_2C also gave a small amount (about 3% yield) of a byproduct identified as the pentacyclic compound 21a (see below).

The fact that dichlorocarbene adds to 7-methyl-6a,7-didehydroglaucone (17b) contrasts with the known C-7 formylation of dehydroaporphines⁷ and suggested the possibility of preparing a simple homoporphine by protecting the C-7 position of the dehydroaporphine. We reasoned that a chlorine or a bromine would be ideal for this purpose because it could be removed at a later stage together with the chlorine or the dichlorocyclopropane intermediate necessary for the ring C homologation. The required 7-halo-6a,7-didehydroaporphine was prepared from the corresponding 6a,7-didehydroaporphine by reaction with NCS or NBS.¹⁵ Thus, when 6a,7-didehydroglaucone (17a) was treated with NBS (benzene; 25 °C; 20 min), we obtained a low yield (26%) of 7-bromo-6a,7-didehydroglaucone (17c), which rapidly decomposed upon manipulation. Nevertheless, 7-chloro-6a,7-didehydroglaucone (17d) could be prepared in 91% yield

(14) Unpublished results.

(15) Huang, S. J.; Lessard, M. V. *J. Am. Chem. Soc.* 1968, 90, 2432.

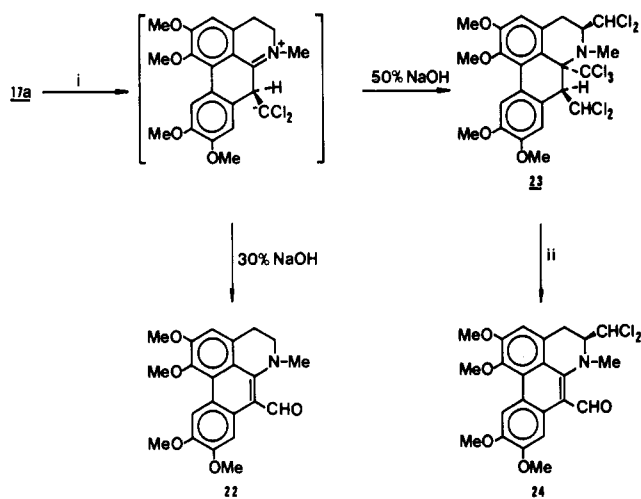
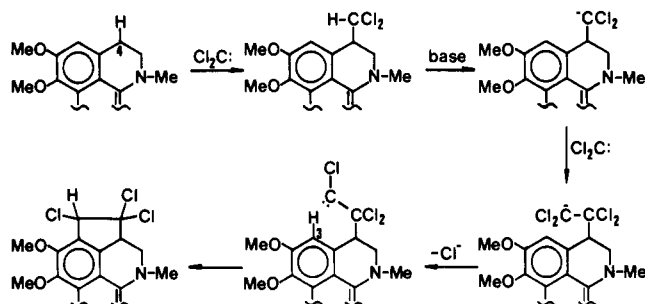


Figure 6. Reagents and conditions: (i) CHCl_3 , 50% or 30% NaOH, TEBA; (ii) 90% EtOH, reflux.

Scheme I



when 17a was reacted with NCS (benzene; 25 °C; 3.5 h), and this product can be stored for long periods of time without decomposition (Figure 3). Unexpectedly, when 17d was reacted with dichlorocarbene (PTM), we did not obtain the desired adduct 18b, but instead a 30% yield of the pentacyclic compound 21b, whose structure was deduced from its ^1H NMR and MS spectra and NOE experiments (Figure 5). A similar product, 21a, was obtained in very low yield from 17b (Figure 3). The formation of 21a and 21b presumably starts with the insertion of dichlorocarbene into the benzylic C-4-H bond and proceeds as shown in Scheme I.

The above results clearly show that the presence of a substituent on the C-7 carbon of the dehydroaporphine substantially modifies the reactivity of the double bond toward dichlorocarbene. Thus, enamine 17b reacts as an olefin to produce adduct 18a, whereas no such adduct is obtained from 17d, which gives instead the insertion product 21b. Surprisingly, when 6a,7-didehydroglaucaine (17a) was treated as above (CHCl_3 , 50% NaOH, TEBA), no 7-formyl derivative 22 was obtained in spite of a published report,⁷ but instead, a 34% yield of 23 was obtained. The formation of 23 can be easily explained as the result of the trapping of the intermediate iminium ion by Cl_3C^- followed by dichlorocarbene insertion (Figure 6). In fact, when the trichloromethyl anion concentration is lowered (30% NaOH instead of 50%), no such trapping occurs and the dehydroaporphine 17a is converted into the aldehyde 22.¹⁶ Further treatment of 22 with CHCl_3 /50% NaOH/TEBA gave a complex mixture of compounds. The highly chlorinated product 23 could be smoothly transformed into the aldehyde 24 by hydrolysis. NOE experiments on 23 point to the stereochemistry shown in Figure

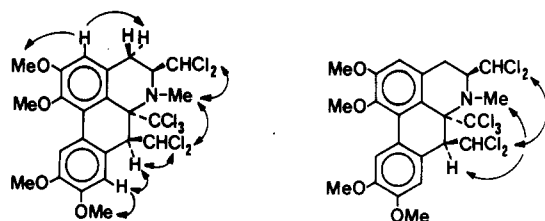


Figure 7. Results of NOE experiments on 23.

7 and, in particular, demonstrate a cis relationship between the two dichloromethyl groups.

Experimental Section

General Procedures. Melting points are uncorrected. NMR spectra were recorded in a Bruker WM 250 spectrometer (250.13 MHz for ^1H , 62.83 MHz for ^{13}C). Chemical shifts are reported in δ with tetramethylsilane as internal reference. Nuclear Overhauser effect difference spectroscopy (NOES) experiments were run with standard Bruker microprograms. THF and toluene were dried from sodium benzophenone ketyl prior to use. Pyridine was distilled from potassium hydroxide and stored over solid KOH. Ethanol-free chloroform was obtained by filtration through neutral alumina (activity I) and stored in the dark under argon. Preparative thick-layer chromatography (preparative TLC) was performed on 20 \times 20 cm plates coated with a 1–2-mm layer of Merck silica gel 60 GF 254. Merck silica gel 60 (70–230 mesh) and Woelm neutral alumina (activity III) were used for column chromatography.

Reaction of N-Carboethoxy-6a,7-didehydronordicentrine (1) with Dichlorocarbene. A solution of 1¹⁰ (100 mg) in chloroform (30 mL) was vigorously stirred with 50% aqueous NaOH (10 mL). After 15 min, tetrabutylammonium chloride (TBAC, 10 mg) was added, and the stirring was continued for 20 min. The organic phase was decanted off and the aqueous layer extracted with chloroform (2 \times 15 mL). The combined organic extracts were washed with water (2 \times 10 mL) and dried over Na_2SO_4 , and the solvent was removed. The remaining material was purified by preparative TLC (silica gel, CH_2Cl_2 -2% 2-propanol), giving 103 mg (86%) of adduct 2, which crystallized from absolute ethanol: mp 160–162 °C; ^1H NMR (80 MHz) δ 8.11 (s, 1 H, Ar), 6.98 (s, 1 H, Ar), 6.60 (s, 1 H, Ar), 6.05 (br s, 2 H, OCH_2O), 3.93 (s, 3 H, CH_3O), 3.87 (s, 3 H, CH_3O), 4.65–2.60 (m, 7 H, CH_2), 1.13 (t, 3 H, $J = 6.3$ Hz, CH_3); ^{13}C NMR δ 156.05, 148.81, 148.75, 147.80, 142.09, 128.92, 122.24, 120.63, 119.04, 116.09, 112.66, 110.89, 107.82, 100.75, 77.20, 64.08, 61.65, 55.93, 55.89, 49.52, 43.35, 27.37, 14.34; IR (KBr) 1700, 1605, 1130, 1100, 1050 cm^{-1} ; MS, m/e 479 ($\text{M}^+ + 2, 2.9$), 477 ($\text{M}^+ + 4.3$), 444 (37), 442 (100), 408 (16), 395 (11), 378 (20). Anal. Calcd for $\text{C}_{23}\text{H}_{21}\text{Cl}_2\text{NO}_6$: C, 57.75; H, 4.43; N, 2.93. Found: C, 57.69; H, 4.47; N, 2.63.

Reduction of Adduct 2 with LiAlH_4 . A solution of adduct 2 (80 mg, 0.17 mmol) in dried THF (5 mL) was added to a stirred suspension of LiAlH_4 (120 mg, 4.28 mmol) in dried THF (15 mL). The reaction mixture was refluxed under an argon atmosphere for 3.5 h before excess hydride was destroyed by dropwise addition of the minimal amount of water. The organic phase was decanted by centrifugation, and the remaining solid was washed with chloroform (4 \times 15 mL). The organic extracts were dried over Na_2SO_4 , and the solvent was evaporated. The remaining solid was purified by preparative TLC (silica gel, CH_2Cl_2 -2% 2-propanol) to yield 50 mg (77%) of 6 and 2 mg (3%) of 5.

5: ^1H NMR δ 7.33 (s, 1 H, H-12), 6.83 (s, 1 H, H-9), 6.68 (s, 1 H, H-3), 6.51 (dd, 1 H, $J = 10.3, 2.0$ Hz, H-8), 6.13 (dd, 1 H, $J = 10.3, 5.1$ Hz, H-7), 6.05 (d, 1 H, $J = 1.4$ Hz, OCH_2O), 5.78 (d, 1 H, $J = 1.4$ Hz, OCH_2O), 3.94 (s, 6 H, 2 CH_3O), 3.58 (dd, 1 H, $J = 5.1, 1.9$ Hz, H-6a), 3.20–2.70 (m, 4 H, H-4, H-5), 2.53 (s, 3 H, NCH_3); MS, m/e 351 ($\text{M}^+ + 82$), 350 (100), 336 (35), 322 (12), 320 (12), 308 (53), 393 (18), 277 (15), 235 (35).

6: mp 191–192 °C (absolute ethanol); ^1H NMR δ 7.27 (s, 1 H, H-12), 6.80 (s, 1 H, H-9), 6.71 (br s, 2 H, H-3, H-8), 6.08 (d, 1 H, $J = 1.4$ Hz, OCH_2O), 5.80 (d, 1 H, $J = 1.4$ Hz, OCH_2O), 3.94 (s, 3 H, CH_3O), 3.93 (s, 3 H, CH_3O), 3.68 (br s, 1 H, H-6a) 3.35 (m, 1 H, H-5), 3.01 (m, 1 H, H-5), 2.69 (m, 2 H, H-4), 2.50 (s, 3 H, NCH_3); ^{13}C NMR δ 148.40, 147.14, 145.66, 143.50, 136.66, 131.77, 128.24, 127.20, 126.08, 124.62, 118.95, 113.24, 110.73, 107.88, 100.51,

(16) Guitián, E., personal communication.

62.11, 55.96, 55.85, 49.54, 44.05, 29.27; IR (KBr) 2980, 2910, 2820, 1600, 1515, 1250, 1100, 1040, 1020, 930, 860 cm^{-1} ; MS, m/e 387 ($M^+ + 2$, 6), 385 (M^+ , 20), 384 (11), 370 (3), 357 (6), 355 (13), 350 (100), 349 (41), 334 (26), 322 (11), 318 (13), 306 (15), 391 (7), 277 (9). Anal. Calcd for $\text{C}_{21}\text{H}_{20}\text{ClNO}_4$: C, 65.37; H, 5.23; N, 3.63. Found: C, 65.44; H, 5.12; N, 3.68.

Homodirectrine (7). To a solution of 7-chloro-7,8-didehydrohomodirectrine (**6**; 35 mg, 0.09 mmol) in absolute ethanol (25 mL) was added anhydrous sodium acetate (45 mg), and the resulting mixture was hydrogenated at atmospheric pressure and room temperature on 5% Pd-C (10 mg). After 16 h, the catalyst was filtered off and the solvent was removed. The residue was dissolved in water (10 mL) and extracted with chloroform (4 \times 20 mL). The solvent was dried (Na_2SO_4) and evaporated to give 30 mg (94%) of homodirectrine (**7**) as an amorphous solid: ^1H NMR δ 7.05 (s, 1 H, Ar), 6.79 (s, 1 H, Ar), 6.59 (s, 1 H, Ar), 6.01 (d, 1 H, $J = 1.4$ Hz, OCH_2O), 5.85 (d, 1 H, $J = 1.4$ Hz, OCH_2O), 3.92 and 3.89 (s, 6 H, 2 CH_3O), 3.36 (dd, 1 H, $J = 11.3$, 6.9 Hz, H-6a), 3.25–2.93 (m, 2 H, CH_2), 2.77 (dd, 1 H, $J = 11.9$, 5.3 Hz, CH_2), 2.61–2.10 (m, 5 H, CH_2), 2.37 (s, 3 H, NCH_3); ^{13}C NMR δ 148.53, 147.14, 145.97, 142.16, 132.52, 128.17, 126.83, 124.60, 120.74, 112.91, 111.76, 107.37, 100.47, 58.34, 56.16, 55.91, 44.86, 41.71, 35.53, 30.32, 25.81; MS, m/e 353 (M^+ , 45), 352 (26), 338 (13), 323 (61), 322 (100), 310(24), 308 (27), 306 (11), 294 (8), 280 (10), 216 (27), 190 (37).

Methiodide of 7: mp 257–259 $^\circ\text{C}$ (acetone); IR (KBr) 1610, 1515, 1465, 1250, 1205, 1100 cm^{-1} . Anal. Calcd for $\text{C}_{22}\text{H}_{26}\text{INO}_4$: C, 53.34; H, 5.29; N, 2.83. Found: C, 53.57; H, 5.06; N, 2.71.

1-[2-[N-Methyl-N-(trifluoroacetyl)amino]ethyl]-3,4,6,7-tetramethoxyphenanthrene (9). To an ice-cold solution of glaucine (**8**; 1.05 g, 2.96 mmol) in pyridine (30 mL), was slowly added trifluoroacetic anhydride (1 mL), and the resulting mixture was heated at 100 $^\circ\text{C}$ for 1 h 20 min under an argon atmosphere. The solvent was removed, and the residue was dissolved in CH_2Cl_2 (50 mL), washed with 10% HCl (4 \times 15 mL), and dried (Na_2SO_4). Evaporation of the solvent gave a solid, which was purified by column chromatography (silica gel, CH_2Cl_2) and crystallized (ether–10% CH_2Cl_2) to give 1.15 g (86%) of **9** as needles: mp 144–146 $^\circ\text{C}$; ^1H NMR δ 9.28 (s, 1 H, H-5), 7.86 and 7.67 (d, 1 H, $J = 9.1$ Hz, H-10, two rotamers, ratio 3:1), 7.60 (d, 1 H, $J = 9.1$ Hz, H-9), 7.23 (s, 1 H, H-8), 7.15 and 7.13 (s, 1 H, H-2, two rotamers), 4.08, 4.06, 4.02, and 3.93 (s, 12 H, 4 CH_3O), 3.76 (t, 2 H, $J = 7.9$ Hz, NCH_2), 3.39 (t, 2 H, $J = 7.9$ Hz, CH_2), 3.12 and 2.92 (s, 3 H, NCH_3 , two rotamers, ratio 1:3); IR (KBr) 3000, 1700, 1595, 1255, 1240, 1110, 990, 870, 845 cm^{-1} ; MS, m/e 451 (M^+ , 52), 324 (8), 312 (23), 311 (100), 281 (6), 277 (9), 265 (17), 140 (23). Anal. Calcd for $\text{C}_{23}\text{H}_{24}\text{F}_3\text{NO}_5$: C, 61.19; H, 5.36; N, 3.10. Found: C, 61.16; H, 5.35; N, 3.34.

Reaction of 9 with Dichlorocarbene. Preparation of 10. A solution of **9** (500 mg, 1.11 mmol) in chloroform (50 mL) was vigorously stirred at 55 $^\circ\text{C}$ with 50% aqueous NaOH (15 mL) and TBAC (20 mg). After 2.5 h, water was added (30 mL) and the organic phase was decanted. The organic layer was extracted twice with CH_2Cl_2 (25 mL), and the combined organic phases were washed once with 10% HCl (15 mL), dried (Na_2SO_4), and concentrated. The residue was purified by column chromatography (silica gel, CH_2Cl_2) to give 410 mg (69%) of adduct **10**, which was crystallized from ether–20% CH_2Cl_2 : mp 145–146 $^\circ\text{C}$; ^1H NMR δ 8.76 (s, 1 H, H-5), 6.94 (s, 1 H, H-8), 6.76 and 6.72 (s, 1 H, H-2, two rotamers), 3.99, 3.93, 3.91, and 3.74 (s, 12 H, 4 CH_3O), 3.69 (m, 2 H, NCH_2), 3.36 (m, 2 H, H-9, H-10), 3.19 and 3.11 (s, 3 H, NCH_3 , two rotamers), 3.07 (m, 2 H, CH_2); IR (KBr) 3140, 3100, 3020, 2980, 1700, 1570, 1470, 1255, 1180, 1140, 1110, 1090, 1045, 990 cm^{-1} ; MS, m/e 535 ($M^+ + 2$, 7), 533 (M^+ , 11), 501 (32), 500 (54), 499 (92), 498 (100), 468 (8), 464 (10), 451 (39), 372 (28), 359 (93), 301 (87), 140 (76). Anal. Calcd for $\text{C}_{24}\text{H}_{24}\text{Cl}_2\text{F}_3\text{NO}_5$: C, 53.94; H, 4.53; N, 2.62. Found: C, 53.96; H, 4.33; N, 2.51.

1-[2-(Methylamino)ethyl]-9,10-(dichloromethano)-3,4,6,7-tetramethoxy-9,10-dihydrophenanthrene (11). To a solution of **10** (270 mg, 0.51 mmol) in methanol (30 mL) and CH_2Cl_2 (5 mL) was added a saturated aqueous solution of Na_2CO_3 (2 mL), and the resulting mixture was stirred at room temperature for 7.5 h. Water (15 mL) was added, and the methanol was removed under reduced pressure (bath temperature 35–40 $^\circ\text{C}$). The aqueous layer was extracted with chloroform (4 \times 15 mL), and the organic extracts were dried (Na_2SO_4) and carefully concen-

trated (35–40 $^\circ\text{C}$) to give the amine **11**, which was used without further purification: ^1H NMR δ 8.76 (s, 1 H, H-5), 6.93 (s, 1 H, H-8), 6.81 (s, 1 H, H-2), 3.98 (s, 3 H, CH_3O), 3.92 (s, 6 H, 2 CH_3O), 3.73 (s, 3 H, CH_3O), 3.36 and 3.29 (AB system, 2 H, $J = 10.9$ Hz, H-9, H-10), 3.00 (m, 4 H, CH_2), 2.53 (s, 3 H, NCH_3).

7-Chloro-7,8-didehydrohomoglaucine (12). A solution of the amine **11** (200 mg) in dried toluene (30 mL) was refluxed under an argon atmosphere for 2 h. The solvent was removed in a rotary evaporator, and the residue was dissolved in chloroform (30 mL), washed with 10% NaOH (1 \times 15 mL), and dried (Na_2SO_4). Evaporation of the solvent followed by preparative TLC purification (silica gel, CH_2Cl_2 –3% EtOH) gave 160 mg (78% yield from **10**) of **12** as a crystalline compound: mp 188–189 $^\circ\text{C}$ (absolute ethanol); ^1H NMR δ 7.34 (s, 1 H, H-12), 6.77 (s, 1 H, Ar), 6.74 (s, 1 H, Ar), 6.68 (d, 1 H, $J = 1.4$ Hz, H-8), 3.94, 3.89, and 3.88 (s, 9 H, 3 CH_3O), 3.63 (d, 1 H, $J = 1.4$ Hz, H-6a), 3.29 (s, 3 H, 1- CH_3O), 3.37–3.25 (m, 1 H, CH_2), 3.13–3.00 (m, 1 H, CH_2), 2.78–2.64 (m, 2 H, CH_2), 2.48 (s, 3 H, NCH_3); ^{13}C NMR δ 151.00 (s), 148.04 (s), 146.57 (s), 145.11 (s), 137.39 (s), 132.45 (s), 129.77 (s), 129.00 (s), 128.17 (s), 127.11 (s), 124.45 (d), 114.44 (d), 111.04 (d), 109.69 (d), 62.06 (d), 60.17 (q), 55.78 (q), 55.74 (q), 55.61 (q), 49.51 (t), 44.10 (q), 29.29 (t); IR (KBr) 2980, 2920, 2830, 2790, 1590, 1515, 1250, 1115, 1020; MS, m/e 403 ($M^+ + 2$, 10), 401 (M^+ , 30), 388 (8), 386 (19), 372 (15), 370 (44), 366 (100), 365 (67), 350 (51), 336 (13), 334 (18), 308 (15), 185 (23), 183 (18). Anal. Calcd for $\text{C}_{22}\text{H}_{24}\text{ClNO}_4$: C, 65.75; H, 6.02; N, 3.49. Found: C, 65.99; H, 5.97; N, 3.31.

Homoglaucine (13). Homoglaucine (**13**) was obtained in 93% yield by catalytic hydrogenation of **12** as in the case of homodirectrine (**7**): ^1H NMR δ 7.10 (s, 1 H, Ar), 6.77 (s, 1 H, Ar), 6.68 (s, 1 H, Ar), 3.94, 3.90, 3.87, and 3.42 (s, 12 H, 4 CH_3O), 3.32–2.96 (m, 3 H, H-6a, CH_2), 2.84–2.74 (m, 1 H, CH_2), 2.68–2.02 (m, 5 H, CH_2), 2.39 (s, 3 H, NCH_3); ^{13}C NMR δ 151.57, 148.28, 146.94, 144.12, 132.46, 131.87, 128.56, 127.00, 126.86, 113.90, 111.20, 110.81, 60.17, 58.32, 55.92, 55.76, 55.71, 45.00, 41.61, 34.90, 29.97, 25.45; MS, m/e 369 (M^+ , 28), 368 (13), 354 (18), 338 (100), 33.6 (5), 322 (10), 232 (9).

Methiodide of 13: mp 234–236 $^\circ\text{C}$ (acetone); IR (KBr) 1590, 1515, 1475, 1330, 1250, 1110, 1095, 1030, 995, 915, 860 cm^{-1} . Anal. Calcd for $\text{C}_{23}\text{H}_{30}\text{INO}_4$: C, 54.01; H, 5.91; N, 2.74. Found: C, 53.84; H, 5.84; N, 2.76.

7,8-Didehydrohomoglaucine (14). To an ice-cold solution of 7-chloro-7,8-didehydrohomoglaucine (**12**; 100 mg, 0.25 mmol) and AlCl_3 (170 mg, 1.20 mmol) in dried THF (40 mL) was added LiAlH_4 (200 mg, 5.27 mmol). The mixture was refluxed under an argon atmosphere for 36 h before excess hydride was destroyed by dropwise addition of the minimal amount of a saturated aqueous solution of Na_2SO_4 . The resulting suspension was centrifuged and the pellet washed with chloroform (3 \times 25 mL). The combined organic extracts were dried (Na_2SO_4) and evaporated to dryness. The residue was purified by preparative TLC (silica gel, CH_2Cl_2 –3.5% EtOH), giving 62 mg (68%) of **14** as an amorphous solid: ^1H NMR δ 7.42 (s, 1 H, H-12), 6.81 (s, 1 H, Ar), 6.74 (s, 1 H, Ar), 6.50 (dd, 1 H, $J = 10.3$, 2.0 Hz, H-8), 6.16 (dd, 1 H, $J = 10.3$, 5.0 Hz, H-7), 3.95, 3.92, and 3.87 (s, 9 H, 3 CH_3O), 3.51 (dd, 1 H, $J = 5.0$, 1.9 Hz, H-6a), 3.26 (s, 3 H, 1- CH_3O), 3.17–2.74 (m, 4 H, CH_2), 2.51 (s, 3 H, NCH_3); ^{13}C NMR δ 150.74 (s), 147.83 (s), 146.26 (s), 145.13 (s), 134.73 (s), 133.32 (s), 129.90 (s), 129.44 (s), 127.27 (s), 126.90 (s), 126.65 (d), 114.69 (d), 111.26 (d), 109.91 (d), 59.94 (q), 59.80 (d), 55.81 (q), 55.74 (q), 55.60 (q), 47.89 (t), 42.62 (q), 28.07 (t); MS, m/e 367 (M^+ , 100), 366 (74), 352 (33), 337 (17), 336 (26), 324 (21), 322 (11), 320 (9), 309 (36), 294 (10), 278 (10), 266 (11), 250 (9).

Methiodide of 14: mp 246–248 $^\circ\text{C}$ dec (acetone); IR (KBr) 1605, 1590, 1515, 1330, 1250, 1120, 1100, 1035, 1005, 990, 920, 870, 810, 780, 760, 750 cm^{-1} . Anal. Calcd for $\text{C}_{23}\text{H}_{28}\text{INO}_4$: C, 54.23; H, 5.54; N, 2.75. Found: C, 54.62; H, 5.67; N, 2.43.

1-[2-[N-Methyl-N-(ethoxycarbonyl)amino]ethyl]-3,4,6,7-tetramethoxyphenanthrene (15). To an ice-cold solution of glaucine (**8**; 330 mg, 0.93 mmol) in pyridine (10 mL) was added dropwise ethyl chloroformate (0.7 mL), and the mixture was heated at 100 $^\circ\text{C}$ for 1.5 h under an argon atmosphere. The reaction mixture was acidified with 10% HCl and extracted with ether (4 \times 25 mL). The ether extracts were dried (Na_2SO_4) and evaporated to give 390 mg (98%) of **15**, which was crystallized from absolute ethanol: mp 109–110 $^\circ\text{C}$; ^1H NMR (80 MHz) δ 9.28

(s, 1 H, H-5), 7.85 and 7.86 (AB system, 2 H, $J = 9$ Hz, H-9, H-10), 7.21 (s, 1 H, H-8), 7.15 (br s, 1 H, H-2), 4.07, 4.04, 4.02, and 3.92 (s, 12 H, 4 CH₃O), 3.68–3.19 (m, 4 H, CH₂), 2.89 (br s, 3 H, NCH₃), 1.25 (t, 3 H, $J = 7.0$ Hz, CH₃); IR (KBr) 3150, 2930, 1685, 1590, 1470, 1305, 1270, 1240, 1120, 890, 865, 845, 770 cm⁻¹; MS, m/e 427 (M⁺, 40), 324 (8), 311 (100), 265 (12), 116 (30). Anal. Calcd for C₂₄H₂₉NO₆: C, 67.43; H, 6.84; N, 3.28. Found: C, 67.86; H, 6.57; N, 3.34.

1-[2-[N-Methyl-N-(ethoxycarbonyl)amino]ethyl]-9,10-(dichloromethano)-3,4,6,7-tetramethoxy-9,10-dihydrophenanthrene (16). The reaction of 15 (200 mg, 0.47 mmol) with dichlorocarbene (PTM; 25 °C; 2 h of stirring) gave, after the usual workup and preparative TLC purification (silica gel, CH₂Cl₂-2% EtOH), 115 mg (48%) of adduct 16, which was crystallized from hexane-ether: mp 143–145 °C; ¹H NMR (80 MHz) δ 8.75 (s, 1 H, H-5), 6.92 (s, 1 H, H-8), 6.77 (br s, 1 H, H-2), 4.17 (q, 2 H, $J = 7.2$ Hz, CH₂O), 3.97 (s, 3 H, CH₃O), 3.92 (s, 6 H, 2 CH₃O), 3.73 (s, 3 H, CH₃O), 2.96 (s, 3 H, NCH₃), 1.28 (t, 3 H, $J = 7.2$ Hz, CH₃); IR (KBr) 3180, 3020, 2930, 1685, 1605, 1590, 1570, 1520, 1470, 1260, 1110, 1050, 1030, 990, 825, 775, 765 cm⁻¹; MS m/e 511 (M⁺ + 2, 5), 509 (M⁺, 7), 487 (13), 486 (29), 485 (34), 484 (63), 427 (7), 359 (23), 343 (13), 323 (16), 311 (21), 115 (100). Anal. Calcd for C₂₅H₂₉Cl₂NO₆: C, 58.83; H, 5.73; N, 2.74. Found: C, 58.77; H, 5.65; N, 2.70.

Reaction of 7-Methyl-6a,7-didehydroglaucaine (17b) with Dichlorocarbene. The reaction of 17b¹⁷ (50 mg, 0.14 mmol) with dichlorocarbene as above (30 min of stirring) gave, after careful solvent evaporation (bath temperature below 40 °C), the unstable adduct 18a as a yellow amorphous solid, which was used without further purification: ¹H NMR δ 8.85 (s, 1 H, H-11), 7.06 (s, 1 H, Ar), 6.71 (s, 1 H, Ar), 3.98, 3.94, 3.93, and 3.74 (s, 12 H, 4 CH₃O), 3.61 (m, 1 H, CH₂), 3.14 (m, 2 H, CH₂), 2.72 (m, 1 H, CH₂), 2.29 (s, 3 H, NCH₃), 1.91 (s, 3 H, CH₃).

Reduction of Adduct 18a with LiAlH₄. Treatment of adduct 18a with LiAlH₄ (60 mg, 0.58 mmol; 3 h of reflux) gave, after the usual workup and preparative TLC purification (silica gel, CH₂Cl₂-1.5% EtOH), 36 mg (63% from 17b; R_f 0.4) of 19a and 2 mg (R_f 0.6) of 21a.

19a: ¹H NMR δ 7.25 (s, 1 H, H-12), 6.94 (s, 1 H, H-9), 6.71 (s, 1 H, H-3), 3.97 (s, 3 H, CH₃O), 3.89 (s, 6 H, 2 CH₃O), 3.63 (s, 1 H, H-6a), 3.40 (s, 3 H, 1-CH₃O), 3.27–3.06 (m, 2 H, CH₂), 2.72–2.62 (m, 2 H, CH₂), 2.45 (s, 3 H, NCH₃), 2.20 (s, 3 H, 8-CH₃); MS, m/e 417 (M⁺ + 2, 15), 415 (M⁺, 41), 402 (10), 400 (27), 386 (18), 384 (45), 380 (65), 379 (88), 364 (100), 350 (12), 348 (20), 336 (35), 322 (14), 320 (13), 306 (17), 278 (10), 192 (27).

Methiodide of 19: mp 204–206 °C (acetone); IR (KBr) 1600, 1590, 1510, 1245, 1115, 1055, 995, 855, 760 cm⁻¹. Anal. Calcd for C₂₄H₂₉ClINO₄: C, 51.67; H, 5.24; N, 2.51. Found: C, 51.43; H, 5.48; N, 2.65.

21a: ¹H NMR δ 7.67 (s, 1 H, Ar), 7.18 (s, 1 H, Ar), 4.80 (s, 1 H, ArCHCl), 4.01 and 3.98 (s, 6 H, 2 CH₃O), 3.76 (dd, 1 H, $J = 15.1$, 8.6 Hz, NCH₂), 3.61 and 3.57 (s, 6 H, 2 CH₃O), 3.24 (dd, 1 H, $J = 15.1$, 7.2 Hz, NCH₂), 2.77 (s, 3 H, NCH₃), 2.62 (s, 3 H, CH₃), 2.30 (dd, 1 H, $J = 8.6$, 7.2 Hz, ArCH); MS, m/e 499, 497, and 495 (M⁺ + 4, M⁺ + 2, and M⁺; relative intensity 33:88:100).

Reduction of Adduct 18a with LiAlD₄. Treatment of 18a with LiAlD₄ (99% deuteration) as above produced 19b in 67% isolated yield with respect to 17b: ¹H NMR (80 MHz) δ 7.24 (s, 1 H, H-12), 6.94 (s, 1 H, H-9), 6.70 (s, 1 H, H-3), 3.96 (s, 3 H, CH₃O), 3.88 (s, 6 H, 2 CH₃O), 3.40 (s, 3 H, 1-CH₃O), 3.40–2.50 (m, 4 H, CH₂), 2.46 (s, 3 H, NCH₃), 2.19 (s, 3 H, 8-CH₃); MS, m/e 418 (M⁺ + 2, 19), 416 (M⁺, 53), 403 (13), 401 (31), 387 (19), 385 (53), 381 (63), 380 (88), 365 (100), 337 (44), 323 (19), 321 (19), 307 (21).

8-Methylhomoglaucine (20a). A solution of 19a (100 mg, 0.24 mmol) in absolute ethanol (25 mL) was hydrogenated at 1 atm on 10% Pd-C for 4.5 h. The usual workup gave 90 mg (98%) of 20a as an amorphous solid: ¹H NMR δ 7.11 (s, 1 H, Ar), 6.71 (s, 1 H, Ar), 6.65 (s, 1 H, Ar), 3.94, 3.89, and 3.86 (s, 9 H, 3 CH₃O), 3.44 (s, 3 H, 1-CH₃O), 3.35 (dd, 1 H, $J = 11.8$, 5.9 Hz, H-6a), 3.17 (dt, 1 H, $J = 11.6$, 4.2 Hz, H-5), 3.08–2.90 (m, 2 H, H-5, H-8), 2.83–2.73 (m, 1 H, H-4), 2.69–2.56 (m, 2 H, H-4, H-7β), 2.40 (s, 3 H, NCH₃), 1.87 (t, 1 H, $J = 12.6$ Hz, H-7α), 0.83 (d, 3 H, $J =$

7.5 Hz, 8-CH₃); ¹³C NMR δ 151.36 (s), 147.75 (s), 146.60 (s), 143.65 (s), 136.57 (s), 132.77 (s), 128.66 (s), 128.19 (s), 125.78 (s), 115.14 (d), 111.90 (d), 111.00 (d), 59.46 (q), 58.01 (d), 55.74 (q), 55.55 (q), 55.50 (q), 45.19 (t), 41.77 (t), 41.74 (q), 36.73 (d), 25.73 (t), 22.24 (q); MS, m/e 383 (M⁺, 22), 382 (12), 268 (24), 352 (100), 340 (12), 336 (8), 246 (22).

Methiodide of 20a: mp 196–198 °C (ethyl acetate); IR (KBr) 1600, 1520, 1470, 1250, 1110, 1070, 1050, 1025, 1010, 970 cm⁻¹. Anal. Calcd for C₂₄H₃₂INO₄: C, 54.86; H, 6.14; N, 2.67. Found: C, 54.98; H, 5.80; N, 2.47.

6a-Deuterio-8-methylhomoglaucine (20b). Catalytic hydrogenation of 19b (40 mg) as above (5 h) gave 35 mg (94%) of the deuterated homoaporphine 20b as an amorphous solid: ¹H NMR δ 7.11 (s, 1 H, Ar), 6.71 (s, 1 H, Ar), 6.65 (s, 1 H, Ar), 3.94, 3.89, and 3.86 (s, 9 H, 3 CH₃O), 3.44 (s, 3 H, 1-CH₃O), 3.16 (dt, 1 H, $J = 11.5$, 4.5 Hz, H-5), 3.05–2.93 (m, 2 H, H-5, H-8), 2.82–2.72 (m, 1 H, H-4), 2.65 (m, 1 H, H-4), 2.58 (dd, 1 H, $J = 12.9$, 9.0 Hz, H-7β), 2.38 (s, 3 H, NCH₃), 1.87 (d, 1 H, $J = 12.9$ Hz, H-7α), 0.82 (d, 3 H, $J = 7.5$ Hz, 8-CH₃); MS, m/e 384 (M⁺, 31), 369 (25), 353 (100), 341 (11), 337 (7), 246 (22).

7-Bromo-6a,7-didehydroglaucaine (17c). To a solution of 17a¹⁸ (50 mg, 0.14 mmol) in dried benzene (15 mL) was added NBS (30 mg, 0.17 mmol), and the resulting mixture was stirred at room temperature and under an argon atmosphere for 15 min. The resulting brown reaction mixture was filtered, and the solvent was removed. The remaining material was purified by preparative TLC (silica gel, CH₂Cl₂-3% EtOH, R_f 0.8), giving 16 mg (26%) of 17c as an unstable solid, which rapidly decomposed on the silica gel plate: ¹H NMR δ 9.31 (s, 1 H, H-11), 7.85 (s, 1 H, H-8), 7.10 (s, 1 H, H-3), 4.10, 4.06, 4.04, and 3.88 (s, 12 H, 4 CH₃O), 3.43 (t, 2 H, $J = 5.7$ Hz, H-5), 3.19 (t, 2 H, $J = 5.7$ Hz, H-4), 2.97 (s, 3 H, NCH₃); MS, m/e 433 (M⁺ + 2, 98), 431 (M⁺, 100), 417 (31), 415 (33), 386 (20), 352 (41), 337 (22), 321 (18), 96 (50), 94 (51), 81 (20), 79 (20).

7-Chloro-6a,7-didehydroglaucaine (17d). To a stirred solution of 17a (200 mg, 0.57 mmol) in dried benzene (20 mL) was added NCS (84 mg, 0.63 mmol), and stirring was continued under an argon atmosphere for 3.5 h. The solvent was removed, and the resulting solid was purified by column chromatography (aluminum oxide activity III, benzene) to give 200 mg (91%) of 17d, which was crystallized from absolute ethanol: mp 142–143 °C; ¹H NMR δ 9.30 (s, 1 H, H-11), 7.78 (s, 1 H, H-8), 7.10 (s, 1 H, H-3), 4.10, 4.06, 4.04, and 3.89 (s, 12 H, 4 CH₃O), 3.44 (t, 2 H, $J = 5.6$ Hz, H-5), 3.19 (t, 2 H, $J = 5.6$ Hz, H-4), 2.99 (s, 3 H, NCH₃); ¹³C NMR δ 150.66 (s), 149.40 (s), 147.76 (s), 144.51 (s), 139.66 (s), 129.95 (s), 126.63 (s), 124.00 (s), 122.12 (s), 120.70 (s), 119.52 (s), 111.93 (d), 108.86 (d), 104.75 (d), 60.02 (q), 56.33 (q), 55.72 (q), 49.63 (t), 41.38 (q), 24.98 (t); IR (KBr) 3140, 2995, 2940, 2845, 1600, 1505, 1250, 1120, 1070, 1010, 840 cm⁻¹; MS, m/e 389 (M⁺ + 2, 37), 387 (M⁺, 100), 374 (13), 372 (37), 356 (7), 353 (12), 314 (13); HR-MS calcd for C₂₁H₂₂ClNO₄ 387.1237, found 387.1235.

Reaction of 7-Chloro-6a,7-didehydroglaucaine (17d) with Dichlorocarbene. To a solution of 17d (50 mg) in chloroform (ethanol free, 20 mL) were added 50% aqueous NaOH (5 mL) and TBAC (5 mg). The two-phase system was stirred at room temperature for 1.5 h. The usual workup followed by column chromatography purification (aluminum oxide activity III, benzene) gave 20 mg (30%) of the pentacyclic compound 21b, which crystallized from methanol-ether: mp 235–238 °C dec; ¹H NMR δ 7.67 (s, 1 H, H-11), 7.55 (s, 1 H, H-8), 4.78 (s, 1 H, ArCHCl), 4.05 (s, 3 H, 9-CH₃O), 3.99 (s, 3 H, 10-CH₃O), 3.81 (dd, 1 H, $J = 15.2$, 8.6 Hz, H-5), 3.62 (s, 3 H, 2-CH₃O), 3.59 (s, 3 H, 1-CH₃O), 3.29 (dd, 1 H, $J = 15.2$, 7.3 Hz, H-5), 2.92 (s, 3 H, NCH₃), 2.35 (dd, 1 H, $J = 8.6$, 7.3 Hz, H-4); IR (KBr) 1620, 1570, 1565, 1510, 1260, 1250, 1110, 1095, 1050, 1020, 795, 780, 760, 750 cm⁻¹; MS, m/e 521, 519, 517, 515 (M⁺ + 6, M⁺ + 4, M⁺ + 2, and M⁺ relative intensity 15:50:100:78); HR-MS calcd for C₂₃H₂₁Cl₄NO₄ 515.0221, found 515.0239.

5,7-Bis(dichloromethyl)-6a-(trichloromethyl)glaucaine (23). A solution of dehydroglaucaine (17a; 60 mg) in chloroform (ethanol free, 30 mL) was vigorously stirred with 50% NaOH (10 mL) and triethylbenzylammonium chloride (TEB, 10 mg) at room temperature for 30 min. The usual workup followed by preparative

TLC purification (silica gel, chloroform) gave 36 mg (34%) of **23** as an unstable solid, which decomposed on heating: ^1H NMR δ 8.17 (s, 1 H, H-11), 6.93 (s, 1 H, H-8), 6.63 (s, 1 H, H-3), 6.49 (d, 1 H, $J = 1.9$ Hz, 7- CHCl_2), 5.77 (d, 1 H, $J = 5.1$ Hz, 5- CHCl_2), 4.64 (d, 1 H, $J = 1.9$ Hz, H-7 α), 3.97 (s, 3 H, 9- CH_3O), 3.94 (s, 3 H, 2- CH_3O), 3.91 (s, 3 H, 10- CH_3O), 3.91 (dd, 1 H, $J = 16.0$, 8.0 Hz, H-4 β), 3.81 (dd, 1 H, $J = 8.0$, 5.1 Hz, H-5 α), 3.66 (s, 3 H, 1- CH_3O), 3.20 (d, 1 H, $J = 16.0$ Hz, H-4 α), 3.11 (s, 3 H, NCH_3); ^{13}C NMR δ 154.61 (s), 149.84 (s), 148.33 (s), 144.31 (s), 131.93 (s), 129.65 (s), 129.01 (s), 121.77 (s), 120.40 (s), 114.83 (d), 111.78 (s), 111.02 (d), 110.51 (d), 77.44 (s), 75.52 (d), 75.32 (d), 67.63 (d) 59.89 (q), 55.97 (q), 55.86 (q), 55.76 (q), 55.40 (d), 39.37 (q), 29.49 (t); IR (KBr) 2940, 2820, 1600, 1575, 1515, 1460, 1395, 1320, 1255, 1215, 1110, 1080, 1025, 980, 880, 870, 800, 780, 770, 735 cm^{-1} .

5-(Dichloromethyl)-7-formyl-6a,7-didehydroglaucaine (**24**).

A solution of **23** (15 mg) in absolute ethanol (6 mL) and water (0.1 mL) was refluxed for 30 min. The solvent was removed, and the residue was purified by preparative TLC (silica gel, chloroform) to give the aldehyde **24** (10 mg, 69%), which crystallized from absolute ethanol: mp 192-194 $^\circ\text{C}$; ^1H NMR δ 10.48 (s, 1 H, CHO), 9.15 (s, 1 H, H-11), 8.95 (s, 1 H, H-8), 7.03 (s, 1 H, H-3), 5.49 (d, 1 H, $J = 9.2$ Hz, 5- CHCl_2), 4.10 (s, 3 H, 9- CH_3O), 4.07 (s, 3 H, 2- CH_3O), 4.04 (s, 3 H, 10- CH_3O) 3.87 (s, 3 H, 1- CH_3O)

3.76-3.62 (m, 2 H, H-4 β , H-5 α), 3.50 (d, 1 H, $J = 15.5$ Hz, H-4 α), 3.49 (s, 3 H, NCH_3); IR (KBr) 2920, 1660, 1590, 1580, 1500, 1460, 1390, 1240, 1120, 1050, 990, 775, 755 cm^{-1} ; MS, m/e 465 ($\text{M}^+ + 2$, 23), 463 ($\text{M}^+ 35$), 437 (8), 435 (13), 381 (26), 380 (100), 364 (20), 352 (60), 336 (17), 334 (10), 322 (12), 306 (9); HR-MS calcd for $\text{C}_{23}\text{H}_{23}\text{Cl}_2\text{NO}_5$, 463.0948, found 463.0952.

Acknowledgment. We thank the Comisión Asesora (CAICYT) for financial support and the Midwest Center for Mass Spectrometry, Department of Chemistry, University of Nebraska, for the high-resolution mass spectra provided. J.L.C. thanks the Ministerio de Educación y Ciencia (Spain) for a grant.

Registry No. 1, 49715-56-2; 2, 96839-01-9; 4, 96839-04-2; 5, 108510-88-9; 6, 96838-54-9; 7, 96838-53-8; 7-MeI, 96838-55-0; 8, 475-81-0; 9, 103769-56-8; 10, 96839-03-1; 12, 96838-49-2; 13, 96838-45-8; 13-MeI, 96838-52-7; 14, 108510-92-5; 14-MeI, 108511-06-4; 15, 103769-55-7; 16, 108510-93-6; 17a, 22212-26-6; 17b, 72498-26-1; 17c, 98931-17-0; 17d, 108510-97-0; 18a, 96839-02-0; 19a, 96838-46-9; 19a-MeI, 108511-03-1; 19b, 96838-47-0; 20a, 96838-50-5; 20a-MeI, 96838-51-6; 20b, 96838-48-1; 21b, 108510-98-1; 23, 108510-99-2; 24, 108511-00-8.

Brominated Tyrosine Metabolites from an Unidentified Sponge

Lili Arabshahi¹ and Francis J. Schmitz*

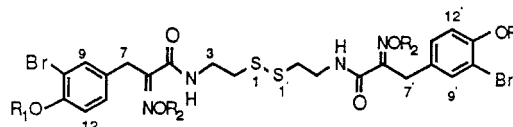
Department of Chemistry, University of Oklahoma, Norman, Oklahoma 73019

Received February 2, 1987

Two new isomeric brominated tyrosine metabolites containing disulfide linkages have been isolated from an unidentified sponge from Guam. Their structures were determined primarily from ^1H and ^{13}C NMR data. (3-Bromo-4-hydroxyphenyl)acetonitrile was also isolated.

Bromotyrosine-derived metabolites are typical of the sponge family Verongidae.² Especially novel among these are the bastadins, some of which are macrocycles comprised of several bromotyrosine units.³ We wish to report novel additions to the bromotyrosine metabolite group, namely, symmetrical compounds with cystamine as the central unit. Among the many brominated tyrosine derivatives, the new compounds **1** and **5** are the first to contain a disulfide moiety. Although the sponge from which these compounds were isolated is unidentified, it is probably of the Verongidae family judging from the nature of its metabolites. Examination of this sponge was prompted by the fact that its extracts showed cytotoxicity, but the compounds described herein are not the cytotoxic components.

The metabolites were obtained from methanol and methanol/chloroform extracts and are quite polar but could be purified by silica gel chromatography followed by reverse-phase HPLC. For the predominant metabolite **1** the molecular formula $\text{C}_{22}\text{H}_{24}\text{Br}_2\text{N}_4\text{O}_6\text{S}_2$, implying 12 degrees of unsaturation, was established by combustion analysis, high-resolution fast atom bombardment (FAB) mass spectrometry, and ^1H and ^{13}C NMR (Table I). Since



	R ₁	R ₂
1 (E,E)	H	H
2 (E,E)	Ac	Ac
3 (E,E)	Me	Me
5 (E,Z)	H	H

only 11 carbon resonances were observed in the ^{13}C NMR spectrum, it could be concluded that **1** was symmetrically dimeric. The presence of amide and possibly oxime or imine groups was indicated by distinct infrared absorptions at 1636 and 1657 cm^{-1} , respectively. ^{13}C NMR signals at 167.5 and 154.7 ppm provided further evidence for these functionalities. Exchangeable proton signals were observed in the ^1H NMR spectrum in pyridine- d_5 at 14 (s), 12 (s), and 8.56 (t) ppm, consistent with the presence of oxime and phenolic protons and the proton of an amide flanked by a methylene group. The amide proton (8.56 ppm) was identified as part of the limited spin system shown in partial structure A by decoupling and deuterium exchange (3.48 ppm, q \rightarrow t). Decoupling confirmed that the three aromatic proton signals were part of a 1,2,4-trisubstituted benzene unit and NOE difference data confirmed that the two-proton singlet at 4.01 ppm was due to a benzylic methylene group situated as shown in partial structure B.

(1) Taken in part from the Ph.D. Dissertation of Lili Arabshahi, University of Oklahoma, 1986.

(2) Faulkner, D. J. *Nat. Prod. Rep.* 1986, 3, 1 and references cited therein. *Marine Natural Products, Chemical and Biological Perspectives*; Scheuer, P. J., Ed.; Academic: New York, 1978-1982; Vol. I-V.

(3) Kazlauskas, R.; Lidgard, R. O.; Murphy, P. T.; Wells, R. J.; Blount, J. F. *Aust. J. Chem.* 1981, 34, 765. Kazlauskas, R.; Lidgard, R. O.; Murphy, P. T.; Wells, R. J. *Tetrahedron Lett.* 1980, 21, 2277.